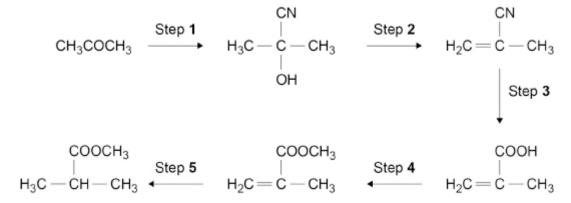
Q1.

Consider this reaction scheme.



Which step is shown with a correct reagent and a correct condition?

- A Step 1 HCN dissolved in water
- B Step 2 KOH dissolved in warm water
- C Step 4 CH₃OH with an alkaline catalyst
- D Step 5 H₂ with a nickel catalyst

(Total 1 mark)

Q2.

Compound **X** can be converted into an alcohol in a two-stage process.

Concentrated Excess
$$\begin{array}{c} H_2SO_4 \\ \hline \end{array} \longrightarrow \text{Intermediate} \begin{array}{c} Excess \\ H_2O \\ \hline \end{array} \longrightarrow \text{Alcohol}$$

What is the name of compound **X**?

- A propene
- B propanal
- C methylbenzene
- **D** ethanamide

(Total 1 mark)